

INFLUENCE OF PRESSURE ON HYDRODYNAMIC INSTABILITY (THERMAL OSCILLATIONS) IN A FORCED FLOW BOILING OF NITROGEN IN A CHANNEL.

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The results of an experimental investigations of conditions of origin and development of hydrodynamic instability - thermal oscillations are presented, at forced flow boiling of nitrogen in a channel in a wide range of pressure variation. In experiences the temperature variations of wall superheat $\Delta T_s = T_w - T_s$ considerable on magnitude and much exceeding a value ΔT_s , applicable to developed nucleate boiling of nitrogen in a channel were captured.

The experience were conducted on the test apparatus described in [1, 2]. In the given work as a test specimen is a tube made of stainless steel (Cr 18%, Ni 10%, Mn 1.5%) with the inner diameter equal to 11.9 mm, length 1447 mm, and wall thickness 0.3 mm was utilized. Orientation of a specimen - vertical.

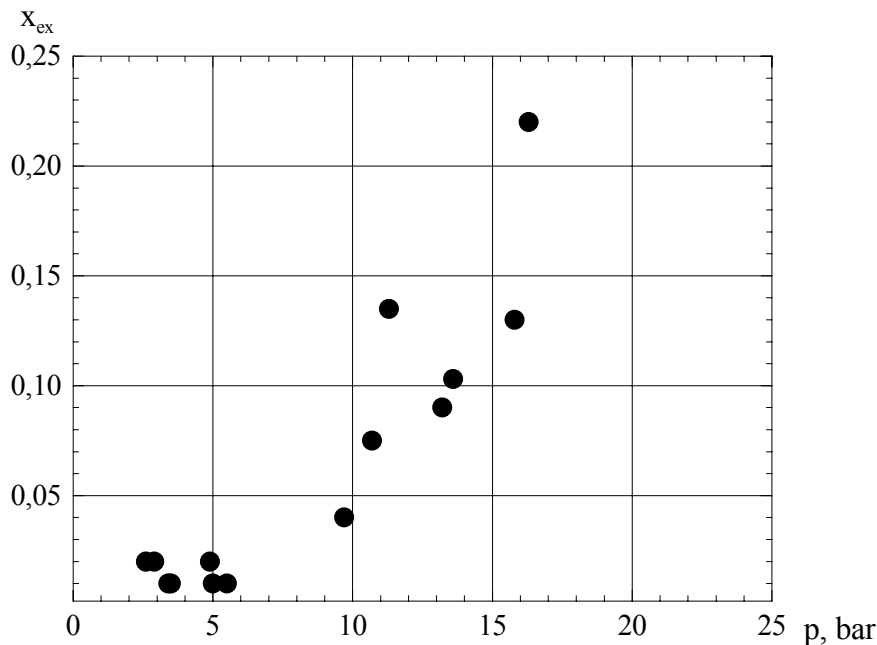


Fig. 1 Exit vapor quality versus pressure
at $G = 340 - 440 \text{ kg}/(\text{m}^2 \text{ sec})$.

The main regime parameters in the experiments varied in the following ranges: mass flow rate $G = 100 \div 700 \text{ kg}/(\text{m}^2 \text{ sec})$; pressure $p = 2,6 \div 16,3 \text{ bar}$; liquid subcooling at a inlet experimental specimen $\Delta T = 2,6 \div 16,9 \text{ K}$; heat flux $q = 0 \div q_{cr} \text{ kW}/\text{m}^2$. In experiences are shown, that with increase of a pressure a nature of oscillations in a forced flow boiling of nitrogen in a channel varies. The thermal oscillations captured in experiences, arose

at positive values vapor quality in output cross-section of a channel and is apparently were are caused by structural modification of a two-phase flow of nitrogen in a heated channel.

A vapor quality in output cross-section of a channel x_{ex} , was calculated on a balance equation:

$$x_{ex} = \frac{i_{in} - i'(p_{in})}{r} + \frac{4l}{d} \cdot \frac{q}{r \cdot G}, \quad (1)$$

As are shown in experiences, exit vapor quality x_{ex} , applicable to beginning of thermal oscillations in a heated channel, is a strong function of value pressure. With increase of a pressure a value exit vapor quality x_{ex} , at which one the thermal oscillations in a channel occur is augmented, see the fig. 1. In paper [3] are shown, that there is a limiting value of pressure, at reaching which one in a forced flow boiling of nitrogen of thermal oscillations disappear. Apparently $x_{ex} = f(p)$ in a fig. 1 asymptotically comes nearer to this value of pressure the graph of relation. Thus increase of pressure in a system appears by the factor stabilizing flow of a boiling nitrogen in a channel.

CONCLUSION

The experimental investigation of thermal oscillations is held at forced flow boiling of nitrogen in a channel. Are shown, that increase of a pressure stabilises flow of a boiling nitrogen. With increase of a pressure the moment of beginning of thermal oscillations displaces in area of higher values exit vapor quality.

NOMENCLATURE

T - temperature,
 i - enthalpy,
 p - pressure
 r - latent heat of vaporization.

Subscripts

w - wall
 s - saturated
 in - input
 ex - exit
 $'$ - corresponds to a fluid

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